



ZYMO RESEARCH

The Beauty of Science is to Make Things Simple

Safety Data Sheet

1 Identification

- **Product:** *TRI Reagent*[®]
- **Cat. Nos:** R2050-1-50, R2050-1-200

- **Supplier:**

Zymo Research Corp.
17062 Murphy Ave.
Irvine, CA 92614
USA
Phone: 1-949-679-1190 or 1-888-882-9682

Manufacturer's MSDS (*TRI Reagent*[®]; Cat. No. TR118) is available at <http://mrcgene.com>.

TRI Reagent[®] is a trademark of Molecular Research Center, Inc.

SAFETY DATA SHEET

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product: TRI Reagent, TRI Reagent LS, TRI Reagent BD
Cat. Nos. TR 118, TS 120, TB 126

Molecular Research Center, Inc.

5645 Montgomery Rd.
Cincinnati, Ohio 45212
USA 1-888-841-0900
Fax: 513-841-0080

Product Name: Tri Reagent, TRI Reagent LS, TRI Reagent BD

Application: Nucleic acid extraction solution for tissue, cells, liquids and blood.

Synonym: Phenol solution

Chemical Formula: A formulation

Molecular Weight: A formulation

CHEMTREC EMERGENCY NUMBER: Only in the event of an emergency involving a spill, leak, fire exposure or accident. USA: 1-800-424-9300; International: 1-703-527-3887

2. HAZARD IDENTIFICATION

Emergency Overview

OSHA Hazards-	Toxic by ingestion, Toxic by inhalation, Toxic by skin absorption, Target Organ Effect, Irritant, Corrosive
Target Organs-	Central nervous system, Liver, Kidney, Pancreas, Spleen
Other Hazards-	Vesicant, rapidly absorbed through the skin
Physical Hazards-	Not hazardous

GHS - Classification

Signal Word: Warning



Health Hazard

Hazard Class	Hazard category	Code	Health Hazard Statements
Acute toxicity, oral	Category 4	H302	Harmful if swallowed
Acute toxicity, dermal	Category 4	H312	Harmful in contact with skin
Skin Corrosion/irritation	Category 1B	H314	Causes skin burns and eye damage
Acute toxicity, inhalation	Category 4	H332	Harmful if inhaled

Precautionary statements

Code	Prevention precautionary statements
P261	Avoid breathing dust/fumes/gas/mist/vapors/spray
P264	Wash...thoroughly after handling
P270	Do not eat, drink or smoke when using this product
P271	Use in a well-ventilated area
P273	Avoid release to the environment
P280	Wear protective gloves/protective clothing/eye protection/face protection
P301+P312	If Swallowed: Call a POISON CENTER or doctor/physician if you feel unwell
P301+P330+P331	If Swallowed: Rinse mouth. Do not induce vomiting
P302+P361+P352	If on skin: Remove/Take off all contaminated clothing. Wash with plenty of soap and water.
P306+P363	If on clothing: Wash contaminated clothing before reuse.
P304+P340	If Inhaled: Remove victim to fresh air and keep at rest in a comfortable position for breathing
P305+P351+P338	If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P303+P361+P353	If on skin or hair: Immediately take off all contaminated clothing. Rinse skin with water/shower.
P309+P311	If exposed or you feel unwell: Call a POISON CENTER or doctor/physician.
P403+P233	Store in well-ventilated location. Keep container tightly closed.

Potential Routes of Exposure and Resulting Health Effects

EYES	Eye contact may be corrosive to tissue, may cause blindness. Acute exposure may result in tearing, conjunctiva swelling, loss of sensation and blurred vision. Chronic exposure or repeated and prolonged exposure to fumes/ vapors may cause corneal ulceration, permanent damage or blindness.
SKIN	Dermal contact may irritate/inflame the skin, with burning sensation or localized loss of feeling (sensitizer, permeator). Skin is a principle route of entry and toxic quantities may be rapidly absorbed. The amount of tissue damage depends on the duration of exposure to the skin. Prolonged exposure can cause severe chemical burns. On skin, there is an initial local anesthesia followed with a white discoloration. Burns may be severe, but painless due to damage to nerve endings. Itching, scaling, reddening and occasionally blistering can characterize skin exposure. Vapors and liquids may be readily absorbed through the skin to cause systemic effects as detailed in acute inhalation exposure. Chronic, long-term exposure may cause dermatitis, and skin sensitization. Pathologic findings include congestion of the lungs, liver, spleen, and kidneys.
INHALATION	Prolonged exposure may cause respiratory tract irritation, injury or arrest. Symptoms of chronic phenol poisoning may include vomiting, difficulty swallowing, diarrhea, anorexia, headache, vertigo, muscle weakness and pain, mental disturbances, dark or smoky urine and possible skin eruptions. Extensive damage to the liver and kidneys may be fatal.
INGESTION	May cause severe burns to the mouth or throat and severe abdominal burning sensation. Ingestion may cause gastrointestinal irritation, nausea, vomiting and diarrhea.

HMIS Classification

NFPA Rating

Health Hazard 3
Chronic Health Hazard *
Flammability 1
Physical hazards 0
PPE = D

Health Hazard 3
Fire 1
Reactivity 0

3. COMPOSITION/Information on Ingredients

Component		Classification	Concentration	
Phenol	CAS-No.	108-95-2	Acute Toxin 3; Skin Corr. 1B	<50 %
	EC-No.	203-632-7		
	Index-No.	604-001-00-2		
Thiocyanate compounds	NA	0022	<30%	
Nonhazardous ingredients	NA	0004	<20%	

4. FIRST AID

SKIN CONTACT: Rescuers should wear protective clothing and gloves while treating patients whose skin is contaminated with phenol. Remove contaminated clothing rapidly and irrigate or wipe exposed areas immediately and repeatedly with low-molecular-weight polyethylene glycol (PEG 300 or PEG 400) which can be diluted to 50% for easier application. Treatment should be continued until there is no detectable odor of TRI Reagent. If PEG is not available, a glycerin solution, olive oil or vegetable oil can be used instead. If these liquids are not available, irrigation with a high-density shower will reduce phenol uptake, but lesser amounts of water will merely dilute the phenol and expand the area of exposure. After treatment with the high pressure shower, the skin should be washed with soap and water for at least 15 minutes. **Decontamination must begin as soon as possible to minimize phenol absorption.** In case of chemical burns, cover area with sterile, dry dressing, bandage securely, but not too tightly. Get medical attention immediately. Double-bag contaminated clothing and personal belongings. See the attached link for additional information (<http://www.atsdr.cdc.gov/MMG/MMG.asp?id=144&tid=27>).

EYE CONTACT: Wash eyes immediately, for at least 15 minutes, with large amounts of water, holding upper and lower lids open. Remove contact lenses, if present and it is easy to do so. Get medical attention immediately.

INGESTION: Wash out mouth if vomiting occurs, have person lean forward with head down to avoid breathing in vomit. Seek immediate medical attention. Do not induce vomiting unless directed to do so by medical personnel. Have conscious person drink several glasses of milk or water. Seek immediate hospital medical attention.

INHALATION: Remove from exposure to fresh air immediately. If breathing has stopped, give artificial respiration. Maintain airway and blood pressure and administer oxygen if available. Treat symptomatically and supportively. Oxygen should be administration by qualified personnel. Get medical attention immediately.

Note to attending physician: No known specific antidote. Areas of skin contact smaller than 100 cm² may cause a minor health hazard. Systemic doses less than 1 gm may cause a minor health hazard although individual sensitivity may vary. For ingestion exposure: give castor oil or other vegetable oil. Give charcoal slurry if conscious. Treat symptomatically. Observe for 24 hrs. Be prepared for emergency cardiovascular intervention. See the following link for additional information (<http://www.cdc.gov/niosh/docs/81-123/pdfs/0493.pdf>).

5. FIRE FIGHTING MEASURES

Moderate fire hazard when exposed to heat or flame. Vapor-air mixtures are explosive above flash point. Vapors are heavier than air and may travel a considerable distance to a source of ignition and flash back. Fires involving phenol should be fought upwind from the maximum distance possible. Emergency personnel should stay away from low areas and ventilate closed spaces before entry.

Flash point: 110° C D93 Method A

EXTINGUISHING MEDIA: Use Class B extinguishers (oils, hydrocarbon liquids.) Dry chemical, carbon dioxide, halon, water spray or standard foam (1987 Emergency Response Guidebook, DOT P 5800.4) for larger fires, use water spray, fog or standard foam (1987 Emergency Response Guidebook, DOT P 5800.4)

FIREFIGHTING: Evacuate area. Wear positive pressure self-contained breathing apparatus. Extinguish using agents indicated. Phenol is combustible and containers may explode in fire. Avoid breathing toxic fumes produced under fire conditions.

6. ACCIDENTAL RELEASE MEASURES

Note that accidental releases may be subject to special reporting requirements and other regulatory mandates. Check and comply with local applicable laws and regulations.

PERSONAL PROTECTIVE EQUIPMENT: Use gloves, boots, Tyvek suit or other impervious covering to avoid skin contact. Use chemical goggles, face shield, or other appropriate eye protection.

SPILL AND LEAK PROCEDURES: Restrict persons not wearing protective equipment from area. Remove all ignition sources. Neutralize spill with slaked lime, sodium bicarbonate or crushed limestone. Collect powdered material and deposit in sealed containers and dispose of phenol as hazardous waste. Isolate area and deny entry.

U.S. DOT EMERGENCY GUIDE # 60

NORTH AMERICAN EMERGENCY RESPONSE GUIDEBOOK # 153

ENVIRONMENTAL PRECAUTIONS: Prevent additional leakage or spillage if safe to do so. Do not let the liquid to enter drains and avoid discharge into the environment.

7. HANDLING AND STORAGE

Observe all Federal, state, and local regulations when storing or disposing of this substance. Store in an area appropriate for flammables; a cool, dry, well-ventilated location, away from direct sunlight, heat or sources of ignition. Avoid contact with hypochlorite, strong oxidizers such as chlorine and bromine.

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

The current OSHA permissible exposure limit (PEL) for phenol is 5 ppm (19 milligrams per cubic meter) as an 8-hour time-weighted average (TWA) concentration. Use engineering controls to keep air borne levels below exposure limit (the human olfactory phenol detection limit is about 5 ppm).

GENERAL PROTECTION AND PRECAUTIONS

PROTECTIVE MEASURES: Do not touch unprotected skin. Do not wear contact lenses while handling this product. **Do not pipette by mouth.** Area ventilation is generally adequate, but use fume hood if available.

AIR PURIFYING RESPIRATOR CANISTERS / CARTRIDGES: Stacked cartridge for organic vapors (black ANSI color code, NIOSH approved) plus dust, mist (red ANSI color code, NIOSH approved).

GLOVES AND PROTECTIVE CLOTHING: User must wear appropriate (impervious) clothing and gloves (rubber or neoprene rubber) to prevent any possibility of skin contact with this substance.

EYE PROTECTION: Safety glasses should be the minimum eye protection. Wear chemical goggles to reduce exposure to aerosols or mists.

EMERGENCY WASH FACILITIES: Where there is any possibility that an employee's eyes and /or skin may be exposed to this reagent; the employer should provide an eye wash fountain and quick drench shower within the immediate work area for emergency use.

ROUTINE OPERATIONS: Lab coats, safety glasses with side shields and gloves should be considered minimum body protection. Wash hands thoroughly after using the reagent and never eat, drink, use tobacco products, apply cosmetics or take medications in areas where a phenol solution is handled, processed or stored. Always wash hands after using the reagent.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical state:	Liquid
Description:	Red to maroon color with a characteristic sweet, medicinal or tar-like odor.
Boiling point:	Not determined
Melting point:	Not applicable
Vapor pressure:	>0.35 mmHg @25C
Evaporation rate:	Not determined
Solvent solubility:	Soluble in water, methanol and glycerol; relatively soluble in aqueous alkali hydroxides, and dimethyl sulfoxide.

10. STABILITY AND REACTIVITY

STABILITY: Stable under normal temperatures and pressures.

INCOMPATIBILITIES: Acetaldehyde: violent reaction.

Aluminum and alloys: may corrode.

Aluminum chloride + nitrobenzene: violent explosion.

1, 3-butadiene, boron trifluoride, and diethyletherate: possible explosion

Calcium hypochlorite: exothermic reaction with possible ignition.

Formaldehyde: possible exothermic reaction.

Lead and alloys: may corrode.

Magnesium and alloys: may corrode.

OXIDIZERS: (strong) Fire and explosion hazard.
Peroxodisulfuric acid: possible explosion.
Peroxomonosulfuric acid: explosion.
Plastics and rubber coatings: may corrode.
Sodium nitrate + trifluoroacetic acid: violent exothermic reaction.
Sodium nitrite: may explode.
Zinc and alloys: may corrode.

DECOMPOSITION: Thermal decomposition products may include toxic oxides of carbon.
Polymerization: Hazardous polymerization has not been reported to occur under normal temperatures and pressures.

CORROSIVITY: Slightly corrosive in the presence of stainless steel. Non-corrosive in glass or polypropylene containers.

11. TOXICOLOGICAL INFORMATION

TOXICITY DATA: Human: (phenol) 10 mg/kg oral-human LDLO. **Rat:** (phenol) 317 mg/kg LD50 oral; (TRI Reagent) 673 mg/kg, acute oral LD50; >1000 mg/kg, acute dermal LD50. (TRI Reagent BD) 534 mg/kg, acute oral LD50. **Mouse:** (phenol) 270 mg/kg LD50 oral. DOT Dermal Toxicity Test, 49CFR 173.137, Class 8, Packing Group II.

Consensus Exposure Guidelines

OSHA permissible exposure limit (PEL): (phenol) 5 ppm (19 mg / m³) as an 8-hour time-weighted average. Skin notation.

NIOSH recommended exposure limit (REL): (phenol) 5 ppm (19 mg / m³) as a time-weighted average for up to a 10-hour workday and a 40-hour workweek. Short-term exposure limit (STEL): 15.6 ppm (60 mg / m³) for periods not to exceed 15 minutes. Skin Notation.

ACGIH TLV: (phenol) 5 ppm (19 mg / m³) as a time-weighted average for up to an 8-hour workday and a 40-hour work week. Skin Notation.

12. ECOLOGICAL INFORMATION

ECOTOXICOLOGICAL INFORMATION: Short-term toxic effects are expected to be limited to the immediate area of environmental release, and will be most pronounced in microorganisms. **Environmental fate:** Expected to rapidly decompose in the environment. **Environmental Movement and Partitioning:** Short-term movement could be due to high water solubility. Movement expected to be limited by relatively rapid environmental detoxification. Phenol is expected to partition strongly into aqueous environmental components.

13. DISPOSAL CONSIDERATIONS

EPA WASTE NUMBER (RCRA HAZARD CLASS) UN1760, 8. All waste disposal activities are subject to federal, state and local laws and regulations. Handle as hazardous waste. Dispose of contents/containers according to local regulations.

14. TRANSPORTATION INFORMATION

Department of Transportation Hazard Classification (DOT) 49CFR172.101; CORROSIVE LIQUID, N.O.S.; UN1760
Department of Transportation Labeling requirements 49CFR172.101; CLASS 8 PACKING GROUP II, PACKING INSTRUCTIONS 851
Department of Transportation Packaging requirements 49CFR173.202; EXCEPTIONS: 49CFR173.154
Shipping designation: (TRI Reagent: guanidine thiocyanate-phenol solution).
ERG Code 153

15. REGULATORY INFORMATION

OSHA: Classified as A HAZARDOUS CHEMICAL@ under US OSHA HAZCOM REGULATION.
TSCA: Some constituents of this product included in US EPA Toxic Substance Control Act (40 CFR part 710).
SARA SECTION 302 Threshold Planning Quantity: 500/10,000 lbs.
CERCLA SECTION 103 Reportable quantity: 1000 lbs.
SARA SECTION 304 Reportable quantity: 1000 lbs.
SARA 311/312 Fire hazard, acute health hazard, chronic health hazard
SUBJECT TO SARA SECTION 313 Annual toxic chemical release reporting.

EUROPEAN UNION:

MAC (GERMANY):	5 PPM PHENOL IN AIR, 19MG/M ³ PHENOL WITH A SKIN WARNING
EINECS#:	2036327
RISK PHASE:	WARNING!
R24/25	TOXIC IN CONTACT WITH SKIN AND IF SWALLOWED.
R34	CAUSES BURNS
SAFETY PHRASE:	
S28	AFTER CONTACT WITH SKIN, WASH IMMEDIATELY WITH PLENTY OF DETERGENT AND WATER.
S45	IN CASE OF ACCIDENT, OR IF YOU FEEL UNWELL, SEEK MEDICAL ADVICE.
SAFETY SYMBOL:	CORROSIVE, TARGET ORGAN TOXICITY, IRRITANT

16. OTHER INFORMATION

Reviewed by	BW, SP
Creation date	9/01/94
Revision date	1/05/15

Reason for Revision: Incorporate Globally Harmonized System of Chemical Classification.

This information is believed to be accurate and represents the information currently available to us. However, we make no warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes.

Additional Information is available at (<http://www.osha.gov/STLC/healthguidelines/phenol/recognition.html>)